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13 States of Matter - CHEMISTRY BATZ KEY IDEA: Temperature and pressure induce changes in the state of substances. The physical state of matter - solid, liquid, gas - is an everyday example of the organization of subatomic particles. Changes in state of matter (freezing, boiling, condensation, etc.) involves a rearrangement of the particles.

13 States of Matter - CHEMISTRY BATZ

Chemistry Chapter 13- States of Matter. STUDY. PLAY. kinetic energy. the energy an object has because of its motion. kinetic theory. a theory explaining the states of matter, based on the concept that all matter consists of tiny particles that are in constant motion. kinetic theory for gases

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Watch different types of molecules form a solid, liquid, or gas. Add or remove heat and watch the phase change. Change the temperature or volume of a container and see a pressure-temperature diagram respond in real time. Relate the interaction potential to the forces between molecules.

States of Matter - Atomic Bonding | Interaction Potential ...

element in the same physical state I. the conversion of a liquid to

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a gas or vapor at a temperature below the boiling point m. states that the tiny particles in all forms of matter are in constant motion n. the change of a solid to a vapor without passing through the liquid state C olumn A amorphous unit cell crystal normal boiling point barometer

STATES OF MATTER 13 - Weebly

Almost all substances can be classified into three states of matter – solids, liquids and gases. Each state has different properties. Heating and cooling a substance can cause it to change state.

States of matter - States of matter - GCSE Chemistry ...

Only the Soviet Union has successfully landed on the Venusian surface—its Venera 13 lander functioned for 127 minutes before succumbing to the elements in 1982.

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