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## **An Activity Series Lab Answers**

Activity Series Lab  
Answers Introduction.  
The purpose of the lab  
was to find which  
metal is the most  
reactive and which  
metal is the least  
reactive. Materials.  
Procedure. Use a  
pipette to fill each of  
the four wells in

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column 1 of the  
reaction plate with 2mL  
of 1.0M aluminum...  
Results. ...

## **Activity Series Lab Answers |**

### **SchoolWorkHelper**

Tag: An Activity Series  
Lab Answers. Activity  
Series Chemistry.

admin July 5, 2018.

Activity Series Today,  
we are going to talk  
about the activity  
series of metals and  
nonmetals, so first

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Lab Answers An  
we're going [...] View  
More Activity Series  
Definition. admin July  
5, 2018.

## **An Activity Series Lab Answers | Activity Series**

If the color of the  
halogen is produced  
there was no reaction.  
From this lab one can  
conclude that the  
activity series for the  
metals, from most  
active to least, is  
Magnesium (Mg), Zinc

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Lab Answers An  
(Zn), Lead (Pb), Copper  
(Cu), Silver (Ag). The  
halogen activity series  
is Chlorine (Cl),  
Bromine (Br), Iodine (I).  
No sources of error.

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Fe 2&plus; will react

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with any ion under it in the activity series that has a charge of  $2+$ .  $\text{Fe}^{3+}$  will only react with ions under it in the activity series with a charge of  $3+$  – Since many transition metals and all Group 2 elements have ions with charges of  $2+$ , while few elements have ions with charges of  $3+$ ,  $\text{Fe}^{2+}$  has many more options for reactions than  $\text{Fe}^{3+}$ .

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## **Activity Series - yaksic.com - high school chemistry lab**

...

The hypothesis has been proven after the experiment. The activity series were correctly found for the metals Magnesium, Zinc, Lead, and Copper and, the halogens Chlorine, Bromine, and Iodine by performing a series of reactions. This lab has almost a

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hundred percent accuracy since all the data found had matched the actual activity series.

## **An Activity Series - WordPress.com**

Activity Series The purpose of this lab was to observe the reactions of metals and the importance of the activity series. Grade and Comments Darbi Jeremy Colleen Class  $\text{KNO}_3$   $\text{Mg}(\text{NO}_3)_2$

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Zn(NO<sub>3</sub>)<sub>2</sub> CuSO<sub>4</sub>

AgNO<sub>3</sub> Distilled water

Copper Iron

Magnesium Tin Zinc

Spot plate scoopula

Students

## **Activity Series Lab Report by Darbi Cook on Prezi**

The activity series is a list of elements in decreasing order of their reactivity. Since metals replace other metals, while nonmetals replace

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other nonmetals, they each have a separate activity series. The table 7.11.1 below is an activity series of most common metals, and the table 7.11.2 is an activity series of the halogens.

## **7.11: The Activity Series - Chemistry LibreTexts**

Oxidation-Reduction Activity Series. Hands-On Labs, Inc. Version 42-0186-00-02. Lab

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Report Assistant. This document is not meant to be a substitute for a formal laboratory report. The Lab Report Assistant is simply a summary of the experiment's questions, diagrams if needed, and data tables that should be addressed in a formal lab report.

**Solved: Please Look  
Over My Lab And Let  
Me Know If My Answ**

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VII. Data Collection:  
Activity Two Go to  
Activity Two and  
repeat what you did in  
sections IV, V, and VI  
with a new set of  
metal/metal ions. Fe<sup>2+</sup>  
Pb<sup>2+</sup> Ni<sup>2+</sup> Sn<sup>2+</sup> Fe No  
reaction Solid plates  
out on the iron strip  
and the color in the  
solution changes from  
colorless to a greenish  
color. Solid plates out  
on the iron strip and



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**Metal/Metal Ion  
Reactions  
Laboratory  
Simulation**

Activity Series of  
Metals Lab. 3 pages.  
Lab 8 Report-1.docx ...

CHEM 101 Tests  
Questions & Answers.

Showing 1 to 6 of 6

View all . Discuss the  
APPLICATIONS of  
LeChatliers Principle in  
daily life (with  
examples) What makes  
the following two  
mixtures different?

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Discuss with an  
example--: 1.

## **CHEM 101 - Barton Community College - Course Hero**

Overview: The students will develop their own activity series of metals based on lab results. Qualitative observations will be used. Tie-ins include single-replacement reactions (SRR), oxidation- reduction, practical applications

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such as galvanization,  
and predicting  
reactions.

### **Developing the Activity Series - ASM International**

The metal activity series indicates the reactivity of metals and how strongly they are reduced. The series begins with lithium, potassium, calcium, and sodium.

### **Metal Activity Series**

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## - **Chemistry | Socratic**

This introductory-level activity can be used as a pre-lab to a unit on Mendelian genetics, and assumes that students are familiar with the terms genotype, phenotype, and allele. [View »](#)

**Dealing with Data In**  
this introductory lab, students collect data and then devise methods to organize and display the data to

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give it more meaning.

## **Carolina LabSheets | Carolina.com**

When the student activity sheet /tutorial is used with computer simulation and the computer animations representing reactions at the particle level (atom level), and when students have the opportunity to do an activity series of metal experiment in the laboratory it is an

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effective way of exposing students to all three levels of representation in Alex Johnstone's triangle: microscopic ...

## **Activity Series of Metals Computer Simulation | Chemdemos**

Grade 11 Chemistry  
Activity Series of Metals Lab Problem:  
What is the order of reactivity of the metals copper, iron,

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## Lab Answers An

magnesium, and zinc in single displacement reactions? Materials: Wellplate/Spotplate Small pieces of magnesium, iron, zinc and copper metal Dilute solutions of hydrochloric acid, copper (II) sulfate, zinc chloride, magnesium chloride, iron (III) sulfate Wash bottle with ...

**Activity Series Lab  
(akey) - Grade 11**

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## Lab Answers Ap **Chemistry Activity ...**

Simple Activity Series  
of Metals Lab for  
Standard Chemistry w  
ww.srhschem.wikispac  
es.com

## **Activity Series of a Metal Lab**

save Save Activity  
Series Lab For Later. 4  
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document as useful 0 0  
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Next. Reactions of  
Copper Experiment 6.  
Activity series lab. AP  
Chemistry - Acid  
Dissociation Constant  
Ka Lab.

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Answers, it is definitely  
simple then, before  
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